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| (21) International Application Number: PCT/GB00/00534 (22) International Filing Date: 18 February 2000 (18.02.00) (30) Priority Data: 9904122.0 23 February 1999 (23.02.99) GB 9911344.1 14 May 1999 (14.05.99) GB (71) Applicant (for all designated States except US): BONAR POLYMERS LIMITED [GB/GB]; Horndale Avenue, Newton Aycliffe, County Durham DL5 6YE (GB). (72) Inventor; and (75) Inventor/Applicant (for US only): MCDONALD, David, John [GB/GB]; 34 Pinewood Crescent, Heighington, County Durham DL5 6RP (GB). (74) Agent: STUTTARD, Garry, Philip; Urquhart-Dykes & Lord, Tower House, Merriion Way, Leeds LS2 8PA (GB). | | (81) Designated States: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i> |
| (54) Title: GLASS COATING (57) Abstract A method for reducing the staining of glass sheets during transportation and storage by applying a coating of a polymer having acid groups. | | |

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GLASS COATING

The present invention relates to the protection of glass and in particular to the protection of sheet glass. More particularly, the present invention relates to the protection of glass surfaces during storage and transportation.

It is known that glass surfaces are vulnerable to staining, in particular when the glass is newly made. Such staining manifests itself as iridescence. Staining is particularly noticeable when glass sheets are stacked side by side so that major surfaces are in adjacent face-to-face relationship. Staining occurs irrespective of whether the major faces are disposed horizontally or vertically or in an intermediate inclined position. The tendency to staining is particularly prevalent in hot humid conditions but is also a significant disadvantage in more temperate climates.

The present invention seeks to provide an improved method for protecting a glass surface by applying a stain-reducing amount of a polymeric solution to combat staining.

Thus viewed from one aspect the present invention provides a method for protecting a glass surface comprising:

forming a solution in a liquid carrier of a polymer having acid groups; and

applying the solution to the glass surface at an application rate of from about 40 to about 500mg of acid groups per square metre of glass surface whereby to form a water soluble coating to reduce iridescence.

Given that the method of the invention is particularly useful in guarding against staining during storage or transportation of a glass article (eg a sheet), it is

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desirable that the solution is subsequently washed off with an aqueous solution (eg water) after storage or transportation.

Preferred acid groups are carboxyl (COOH) and sulphonic (HSO₃) groups. Related carboxyl groups such as acyl substituent groups may also be useful eg anhydrides.

Preferably the rate of application of the polymer is from about 50 to about 500 milligrams of acid groups per square metre of glass surface. In practice, the rate of application will be tailored to the climatic conditions (eg temperature and humidity) in which the glass surface is found. Generally speaking, in tropical conditions, application rates at the upper end of the range are preferred and in more temperate climates, the lower end is preferred.

Typically the rate of application may be measured by washing off the applied coating from a known area of glass (normally with distilled or demineralised water) and titrating against sodium or potassium hydroxide using an indicator (eg phenolphthalein indicator). By way of example, the following calculation may be used for polyacrylic acid:

Equivalent weight of polyacrylic acid = 72g
Equivalent weight of KOH = 56.11g (also = 1000ml of 1M KOH solution or 10000ml of 0.1M KOH solution)
1.0g of PAA = (10000/72) = 138.88ml of 0.1M KOH solution

Thus a solution containing 250mg of PAA will give a titre of 34.7mls of 0.1M KOH (or NaOH) or each mg of PAA will give rise to a titre of 0.1388ml of 0.1M KOH (or 1.388ml of 0.01M KOH). In order to equate this to a treatment level per square metre, the following expression is used:

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Treatment level (in mg/m²) = titre/(area in m² x 0.1388)

Thus a litre of 7ml of 0.1M KOH from a sheet of 0.5m² area would indicate a treatment level of 100.8 mg/m² (and of course a titre of 70ml would be obtained from the same sheet using 0.01M KOH).

In a preferred embodiment the polymer has an acid value of 500mg/g or more, preferably 600mg/g or more. Preferably the acid value is in the range 500-1000mg/g, particularly preferably in the range 600-800mg/g. For example, polyacrylic acid has an acid value of about 760mg/g KOH and polymethacrylic acid has an acid value of about 652mg/g KOH.

The polymers for use in the invention may be homopolymers or heteropolymers. Preferred polymers are polymers of acrylic acid or methacrylic acid. For example, polyacrylic acid or polymethacrylic acid may be used. Sulphonic acid functional derivatives of polyolefins and polystyrene may also be suitable. Where the polymer is a heteropolymer, co-monomers may be selected from the group consisting of acrolein, simple or branched alkyl (eg C₁₋₆-alkyl) or hydroxyesters of acrylic acid or methacrylic acid, vinyl derivatives of benzene or toluene (eg styryl or cinnamyl), vinyl esters of organic acids (eg vinyl C₁₋₆-carboxylates such as vinyl acetate, vinyl butyrate and vinyl propionate) and maleic anhydride.

The polymer may have a molecular weight between preferred upper and lower limits. The choice of these limits will be within the capabilities of the man skilled in the art. Nevertheless, below a molecular weight of about a few hundred there may be little or no film-forming effect. The upper limit of the molecular weight may be as high as several million. However, above a molecular weight of about 600,000,

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the polymer chains may be sufficiently sterically hindered in solution so as to lose their effective carboxyl or other concentration. In the downstream handling of the "glass stack" of which the glass surface is a part, it is desirable that the applied solution is capable of being washed off. This has an effect on the choice of the upper limit of the molecular weight range because as the molecular weight increases the ease of solubilisation decreases. Generally speaking, a preferred molecular weight range is 10,000 to 600,000.

The preferred polymers for use in the method of the invention are polymers of acrylic acid or methacrylic acid (eg polyacrylic acid or polymethacrylic acid) having molecular weights in the range of about 10,000 to about 600,000 and solution viscosities (in water or organic solvent) of about 12 to about 100 centipoises, at 5% polymer concentration and 25°C.

The solution may be applied to the glass surface in any suitable manner. Surfactants may be present to assist wetting of the glass surface. De-foamers or anti-foamers may also be present to suppress the tendency to foam which may otherwise hinder the formation of a coating of the polymer on the glass surface. In addition, dyes may be present. Other optional ingredients may be used provided that they do not reduce the effect of the acid groups in the applied composition.

Typically, the glass surface will be part of a sheet (eg a flat glass or float glass sheet).

Viewed from a further aspect the present invention provides a glass sheet having on at least one surface thereof a coating of a polymer solution having acid groups in sufficient concentration to reduce iridescence. Preferably

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the extent of coverage is 40 to about 500 milligrams of acid groups per square metre of glass surface. Preferably, the glass sheet is obtainable by a method as hereinbefore defined.

The present invention will now be described in a non-limitative sense with reference to the following examples:

Example 1

A formulation comprising:

100g of an aqueous solution of polyacrylic acid (Acrysol A1/Primal A1/Acumer 1510 - Rohm & Haas) at a concentration of 25wt% and a molecular weight of approximately 20,000,
1.0g of silicone oil emulsion (NL65 - Dow Corning),
0.3g non-ionic surfactant (Nonidet LE - Shell) and
0.1g cochineal dye (Carmoisine - Supercook),
2400g water

was sprayed onto a glass sheet to form a continuous functional film which was ultimately easily removable with water. The film reduced the risk of iridescent staining.

Example 2

A formulation comprising:

160g of an aqueous solution of polyacrylic acid (Glascol E7 - Ciba Speciality Chemicals) at a concentration of 25wt% and a molecular weight of approximately 30,000,
1.5g of hydrophobic silica in mineral oils (Contrapen PR/388 - Lamirsa Laboratories Miret SA),
0.3g non-ionic surfactant (Nonidet LE - Shell) and
0.1g cochineal dye (Carmoisine - Supercook),

1350g water

was sprayed onto a glass sheet to form a continuous functional film which was ultimately easily removable with water. The film reduced the risk of iridescent staining.

Example 3

A formulation comprising:

160g of an aqueous solution of polyacrylic acid (Glascol E9 - Ciba Speciality Chemicals) at a concentration of 25wt% and a molecular weight of approximately 75,000,
1.5g of hydrophobic silica in mineral oils (Contrapen PR/388 - Lamirsa Laboratories Miret SA),
0.3g non-ionic surfactant (Nonidet LE - Shell) and
0.1g cochineal dye (Carmoisine - Supercook),
1350g water

was sprayed onto a glass sheet to form a continuous functional film which was ultimately easily removable with water. The film reduced the risk of iridescent staining.

Example 4

A formulation comprising:

160g of an aqueous solution of polyacrylic acid (Glascol E11 - Ciba Speciality Chemicals) at a concentration of 25wt% and a molecular weight of approximately 250,000,
1.5g of hydrophobic silica in mineral oils (Contrapen PR/388 - Lamirsa Laboratories Miret SA),
0.3g non-ionic surfactant (Nonidet LE - Shell) and
0.1g cochineal dye (Carmoisine - Supercook),
1350g water

was sprayed onto a glass sheet to form a continuous functional film which was ultimately easily removable with water. The film reduced the risk of iridescent staining.

Example 5

A formulation comprising:

160g of an aqueous solution of polyacrylic acid (Glascol K11 - Ciba Speciality Chemicals) at a concentration of 25wt% and a molecular weight of approximately 10,000,
1.5g of hydrophobic silica in mineral oils (Rhodoline DF6681 - Rhodia),
0.3g non-ionic surfactant (Nonidet LE - Shell) and
0.1g cochineal dye (Carmoisine - Supercook),
1350g water

was sprayed onto a glass sheet to form a continuous functional film which was ultimately easily removable with water. The film reduced the risk of iridescent staining.

CLAIMS:

1. A method of protecting a glass surface comprising:
forming a solution in a liquid carrier of a polymer having acid groups; and
applying the solution to the glass surface at an application rate of from about 40 to about 500mg of acid groups per square metre of glass surface whereby to form a water soluble coating to reduce iridescence.
2. A method as claimed in claim 1 comprising the step of:
washing the applied solution off the glass surface with an aqueous solution at the end of the desired period.
3. A method as claimed in either of claims 1 or 2 wherein the polymer is a polymer or co-polymer comprising a monomer selected from the group consisting of acrolein, vinyl derivatives of benzene or toluene, vinyl derivatives of organic acids, simple or branched alkyl or hydroxyesters of acrylic acid or methacrylic acid and maleic anhydride.
4. A method as claimed in either of claims 1 or 2 wherein the polymer is a polymer of acrylic acid, methacrylic acid or a sulphonic acid functional derivative of a polyolefin or polystyrene.
5. A method according to claim 4 wherein the polymer is a polymer of acrylic acid or methacrylic acid having a molecular weight in the range of about 10,000 to about 600,000.
6. A method as claimed in claim 5 wherein the polymer is applied in solution having a solution viscosity

(in water or in organic solvent) of about 12 to about 100 centipoises at 5% polymer concentration and 25°C.

7. A method as claimed in any preceding claim wherein the polymer has an acid value of 500mg/g or more.

8. A method as claimed in claim 7 wherein the polymer has an acid value of 600mg/g or more.

9. A method as claimed in claim 7 or 8 wherein the polymer has an acid value in the range 500-1000mg/g.

10. A method as claimed in claim 7, 8 or 9 wherein the polymer has an acid value in the range 600-800mg/g.

11. A glass sheet having on at least one surface thereof a coating of a polymer having acid groups in sufficient concentration to reduce iridescence.

12. A glass sheet as claimed in claim 11 having an extent of coverage being about 40 to about 500mg of acid groups per square metre of glass surface.

13. A glass sheet as claimed in claim 11 obtainable by a method as defined in any of claims 1 to 10.

INTERNATIONAL SEARCH REPORT

International Application No.

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A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C03C17/32 C09D5/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C03C C09D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

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